| Name: | Date: |
|-------|--------------|
| A & P | Ms. Hartnett |
| | |

Chapter 2, Part 2: Intro to Biochemistry

| I. <u>BI</u> | OCHEMISTRY: ESSENTIALS FOR LIFE |
|---------------|---|
| • | compounds: |
| • | Contain |
| • | Most are bonded |
| • | Example: C ₆ H ₁₂ O ₆ () |
| • | compounds: |
| • | carbon |
| • | Tend to be compounds |
| • | Example: H ₂ O (water) |
| | |
| II. <u>IM</u> | PORTANT INORGANIC COMPOUNDS |
| 1 | |
| 2 | |
| | |
| | |
| | |
| Water | |
| • Most _ | inorganic compounds |
| • | of our body |
| • | properties: |
| • | High |
| • | /solvent properties |
| • | Chemical |
| • | |

Salts

• ____ = ___ compounds

• Easily _____ into ____ in the presence of water (_____,

____)

• Vital to many body ______ (nerve ______,

transporting ______)

• Include _____ which conduct electrical currents

Acids and Bases

Acids

• Can release detectable (H⁺) ions

• Proton

Bases

• Releases _____(OH⁻) ions

• Proton _____

• _____reaction

• Acids and bases react to form _____ and a

pН

Measures relative _____ of

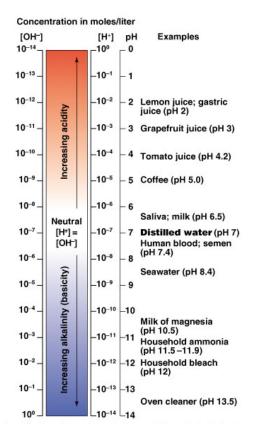
ions

• pH 7 = _____

• pH below 7 = _____

• pH above 7 = _____

• Buffers = chemicals that can _____ pH change



III. IMPORTANT ORGANIC COMPOUNDS

| 1 | _ | |
|---|-------|--|
| 2 | | |
| 3 | | |
| 4 | | |

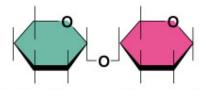
Carbohydrates

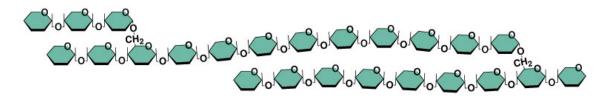
| • | Contain | , | ,, aı | nd oxyge | en |
|---|---------|---|-------|----------|----|
| | | | | | |

- Include _____ and ____
- Classified according to _____
 - ____saccharides simple sugars
 - _____saccharides _____ simple sugars joined by dehydration synthesis
 - ____saccharides long _____chains of ____simple sugars
- Usually end in _____



(a) Simple sugar (monosaccharide) (b) Double sugar (disaccharide)





(c) Starch (polysaccharide)

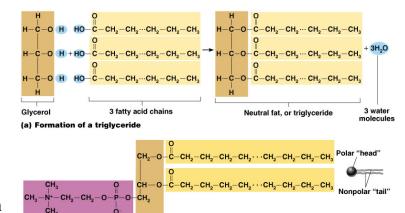
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waxes

- Contain ______, hydrogen, and oxygen
 - Carbon and hydrogen

____oxygen

• _____ in water



2 fatty acid chains (nonpolar end)

H₃C

CH₃

(b) Phospholipid molecule (phosphatidyl choline)

CH₃

Common Lipids in the Human Body

- Neutral fats (______)
 - Found in fat _____
 - Composed of _____ and
 - Source of energy
- _____ = form cell membranes
- Steroids
 - Include _____, bile salts, vitamin D, and some _____

Proteins

- Made of ______
 - Contain ______, oxygen, hydrogen, ______, and sometimes

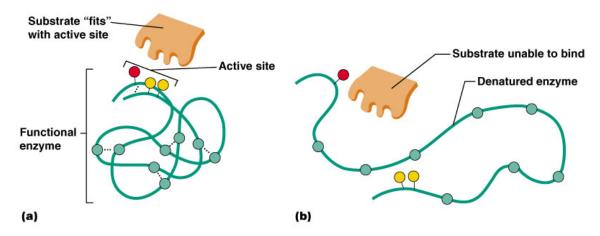
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(c) Cholesterol

- Account for over _____ of the body's organic matter
 - Provides for _____ materials for body tissues
 - Plays a vital role in cell _____
- Act as _____, hormones, and _____

Enzymes

- Act as biological _____
- Increase the _____ of chemical _____
- Anything ending in ______ is an enzyme

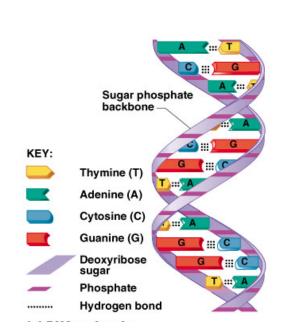


Nucleic Acids

- Provide _____ of life
- Control body ______ by dictating _____ structure
- ______ bases (building blocks)
 - A = Adenine
 - G = Guanine
 - C = Cytosine
 - T = Thymine
 - U = Uracil
- Make _____ and ____

Deoxyribonucleic Acid (DNA)

- Organized by ______ bases to form
 - _____ before cell division



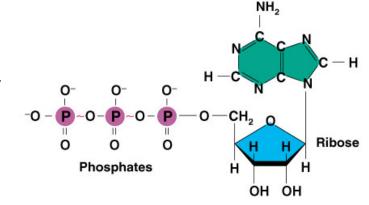
• Provides _____ for every ____ in the body

Ribonucleic Acid (RNA)

- Carries out the ______ for _____ synthesis issued by DNA
- ______ -stranded
- Has _____ instead of thymine
- Made of _____ sugar instead of deoxyribose

IV. ADENOSINE TRIPHOSPHATE (ATP)

- Chemical _____ used by all cells
- Energy is ______ by breaking high
 energy _____ bond
- ATP is replenished by ______ of food fuels



Adenine

(a) Adenosine triphosphate (ATP)

How ATP Drives Cellular Work

